



# Thomas Zemb

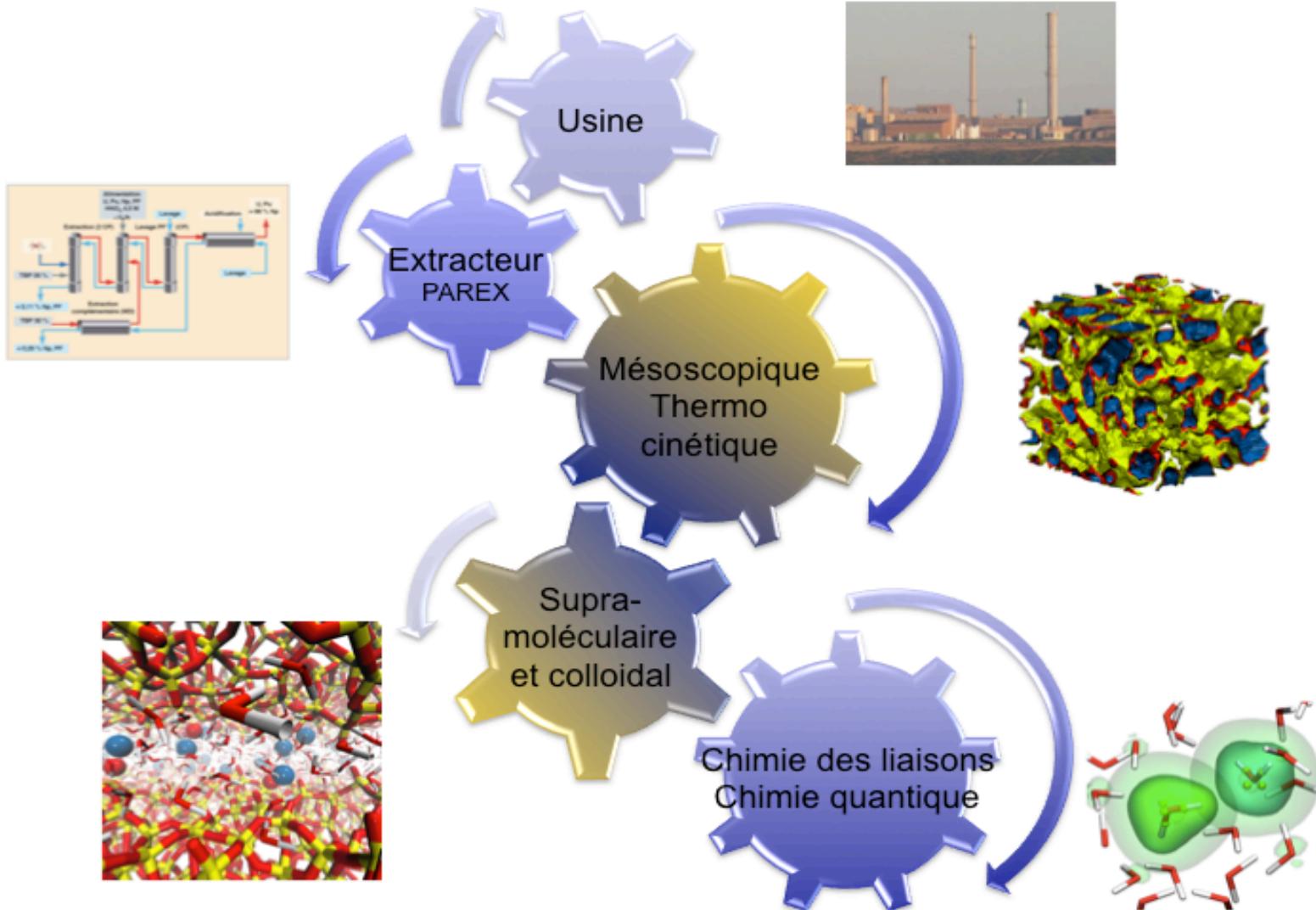
*lecture n°10:  
The chemical potential step as molecular driving  
force - Thomas Zemb*



2014-2015



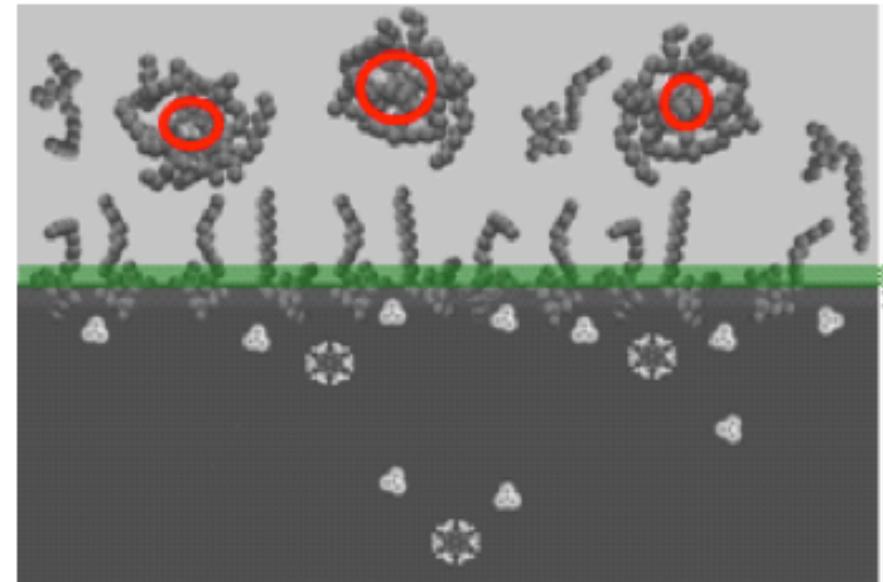
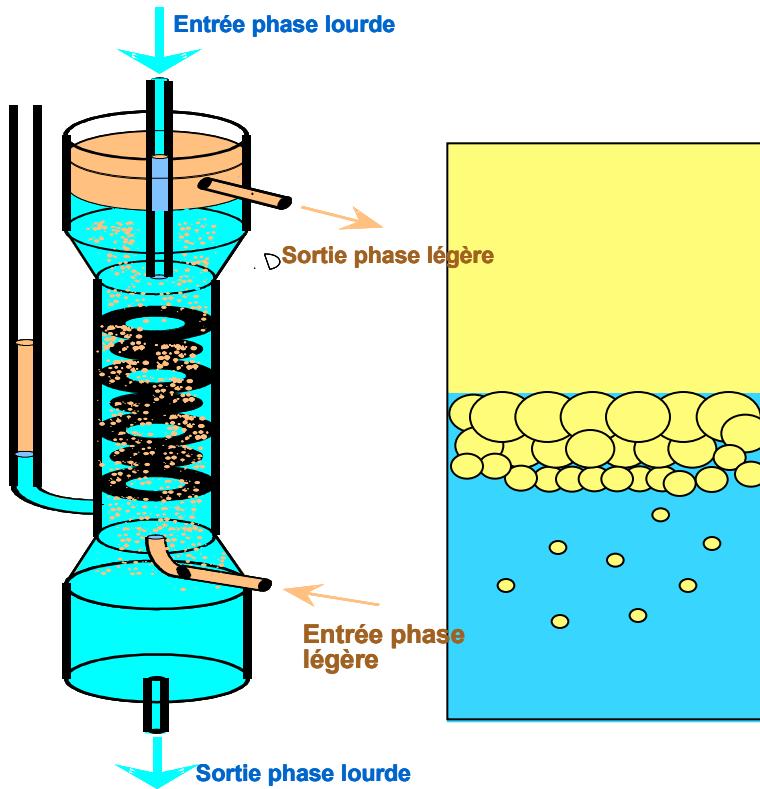
# An intrinsic multi-scale approach :





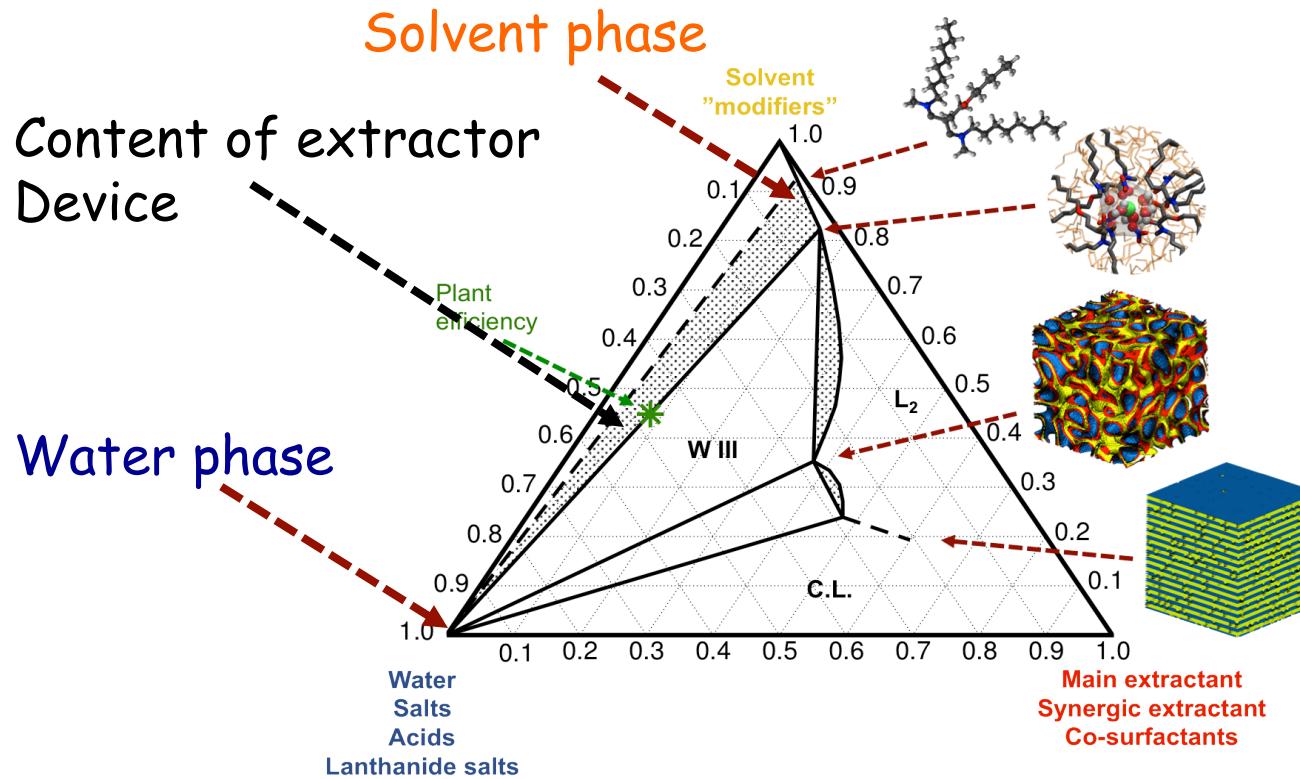
# Content

- **Going down to meso-scale („nano“)**
- Locating w-rich and O-rich in the phase diagramm
- Mapping the three-phase accident
- The chemical potential level representation
- The driving force for phase transfer and selectivity
- Decomposing the chemical potential step in:
  - Complexation (frist neighbour)
  - „Bulk“ terms
  - Curvature, dispersion, entropic terms



« Green » interface :  $\Sigma$  of macroscopic contact

« Red» interface in : extraction as an adsorption isotherm

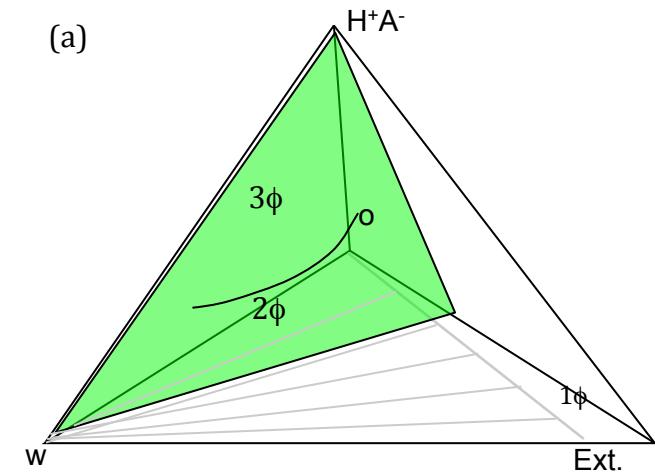


- Y. Chevalier et Th. Z. Reports in Progress in Physics (1991)

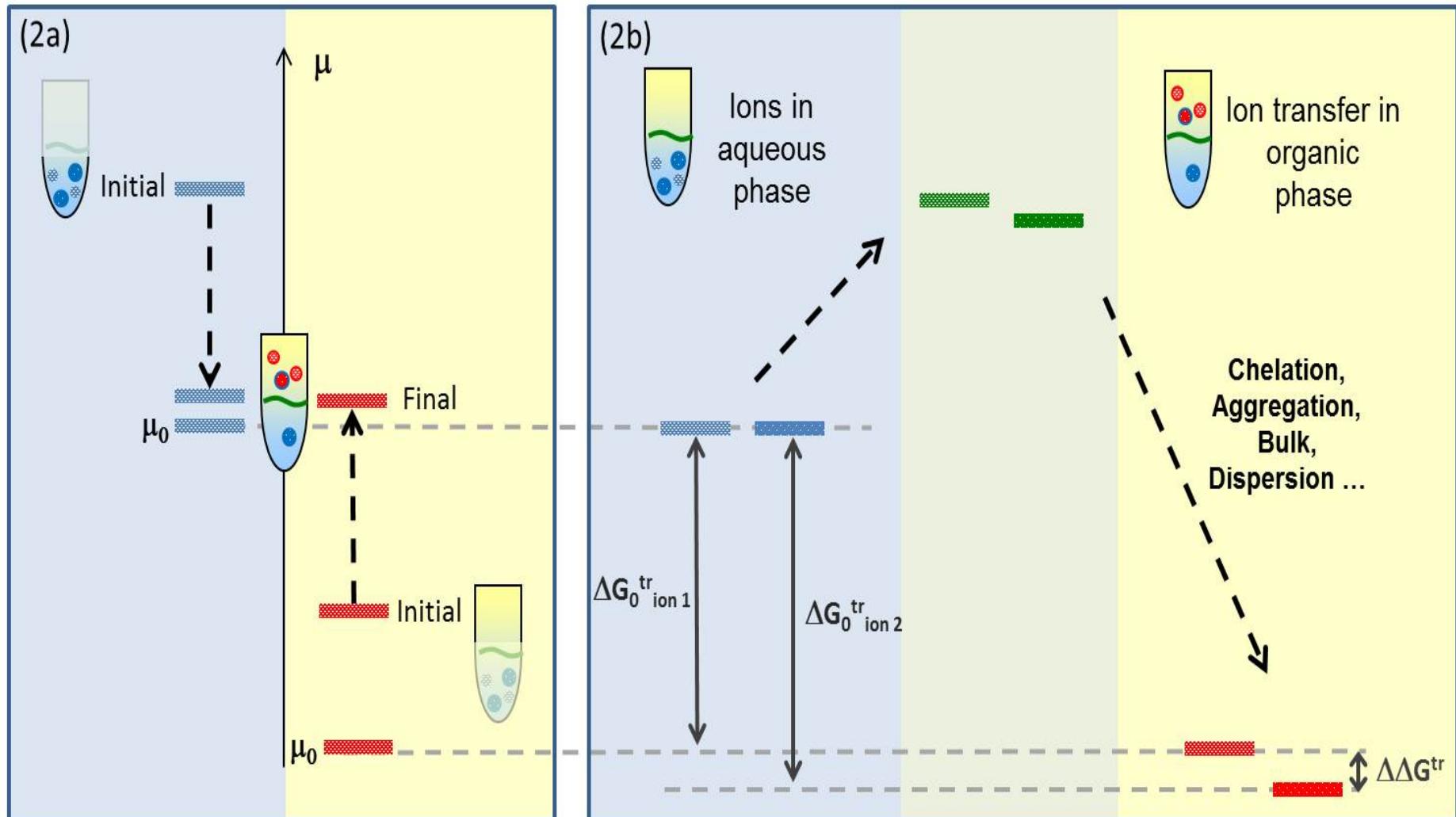
**Distribution coeff.**

$$\mathbf{Y}_f = \mathbf{D} \cdot \mathbf{X}_f$$

$$D_{app} = \frac{\left( x_{C^+ A^-} \right)_{O-rich}}{\left( x_{C^+ A^-} \right)_{W-rich}}$$



Bauer C, Bauduin P et al. Liquid/liquid metal extraction: Phase diagram topology resulting from molecular interactions between extractant, ion, oil and water. Eur Phys J Spec Top 2012;213:225–41.



Zemb T, Bauer C, Bauduin P, Belloni L, Déjugnat C, Diat O, et al. Recycling metals by controlled transfer of ionic species between complex fluids: en route to "ienaiics." Colloid & Polymer Sci 2015:1-22.

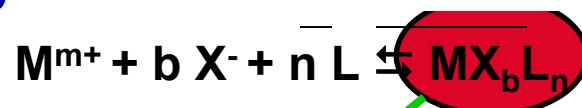


## The driving forces competing for extraction :

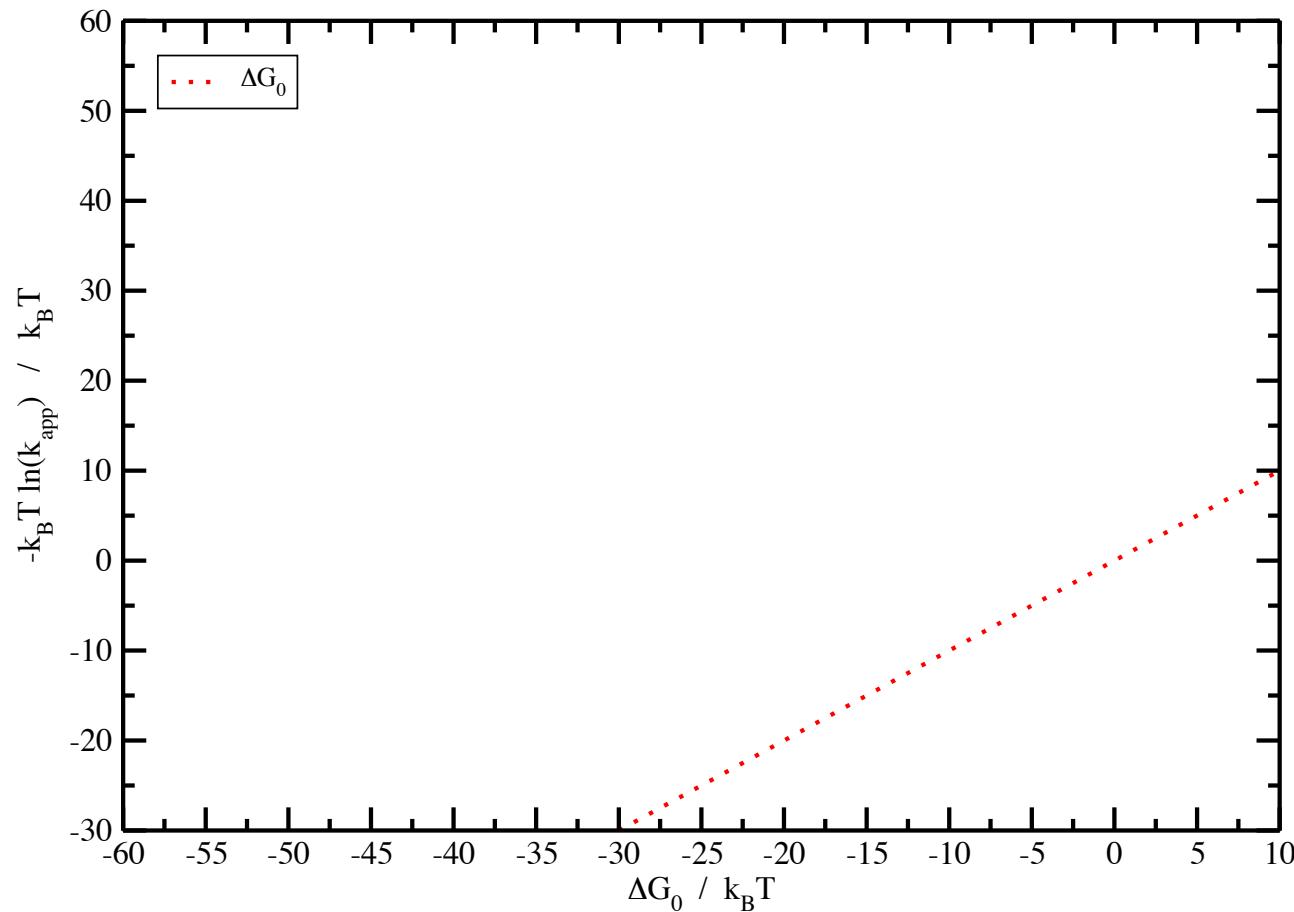
$$\delta\mu_0 = (\mu_0^{o-rich} - \mu_0^{w-rich}) = \delta\mu_{compl(O\_W)} + TdS_{conf} + \delta\mu_{Bulk} + \delta\mu_{D-I-FB}$$

$$n.\delta\mu_0 = dG_0 = kT.\ln K_{app}$$

Link with parametric and activity " corrections" used for mass Conservation in design:

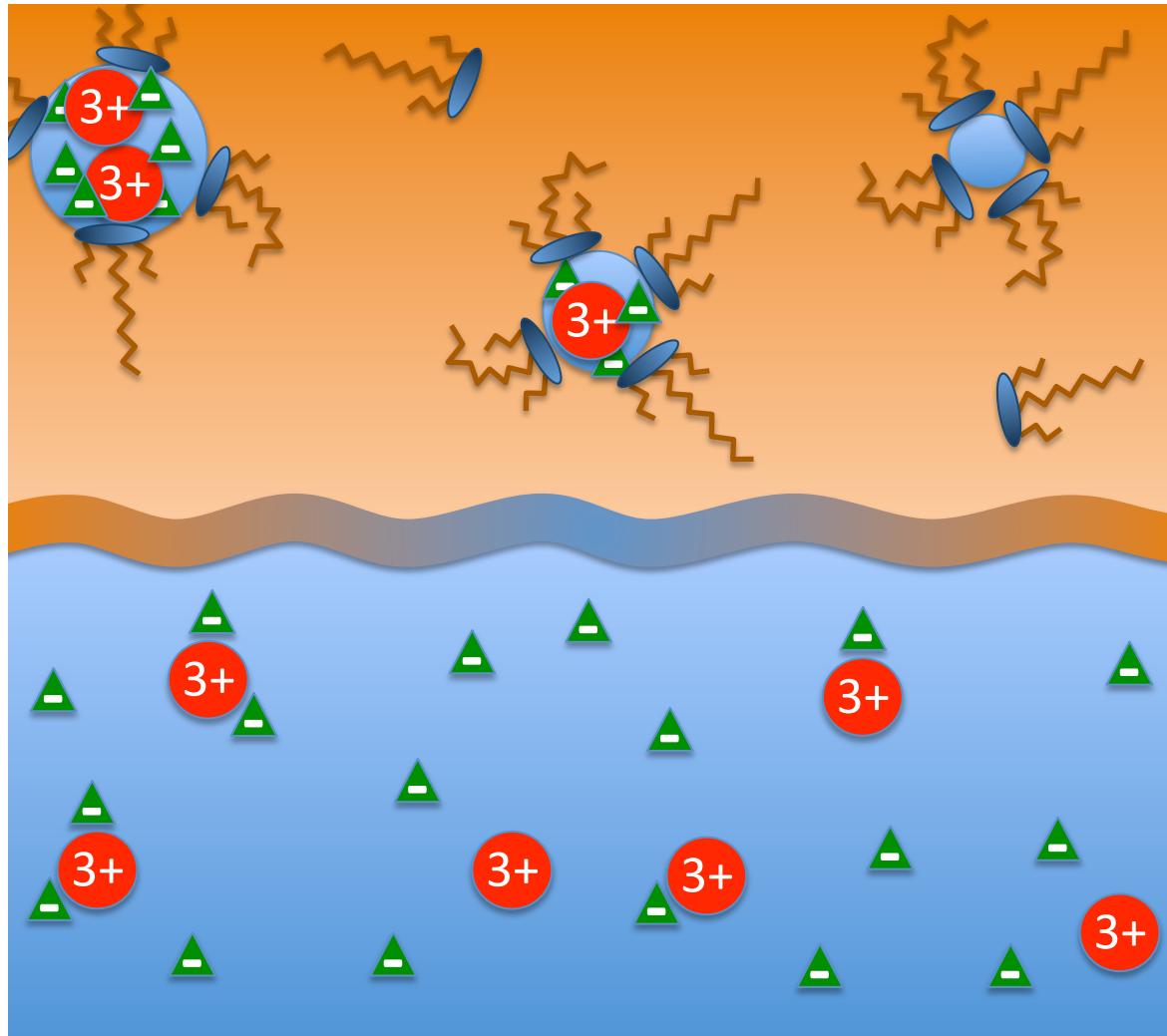


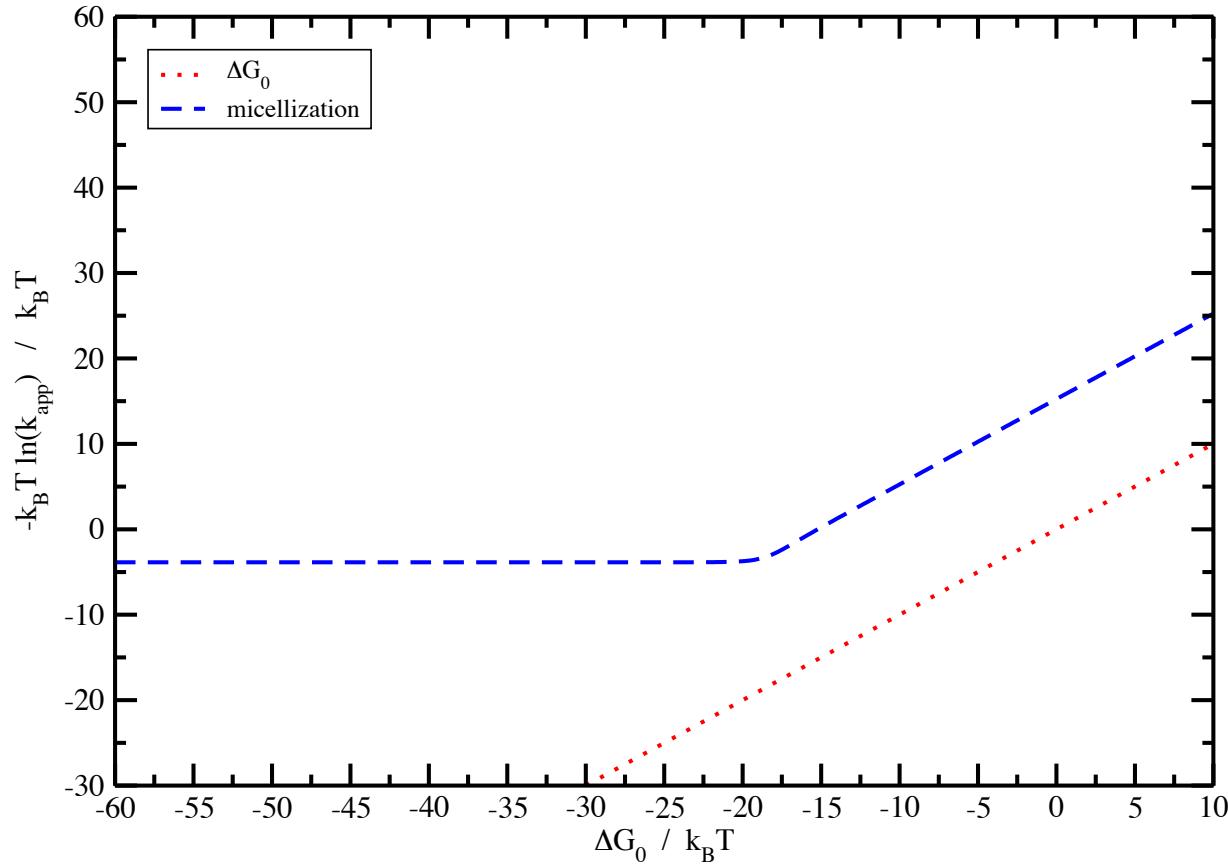
$$c_M^{\text{org}} = [MX_b L_n] = K_{ext}^\circ [M^{m+}] [X^-]^b [L]^n \cdot \frac{\gamma_L^n}{\gamma_{MX_b L_n}} \cdot \gamma_{M^{m+}} \cdot \gamma_{X^-}^b$$



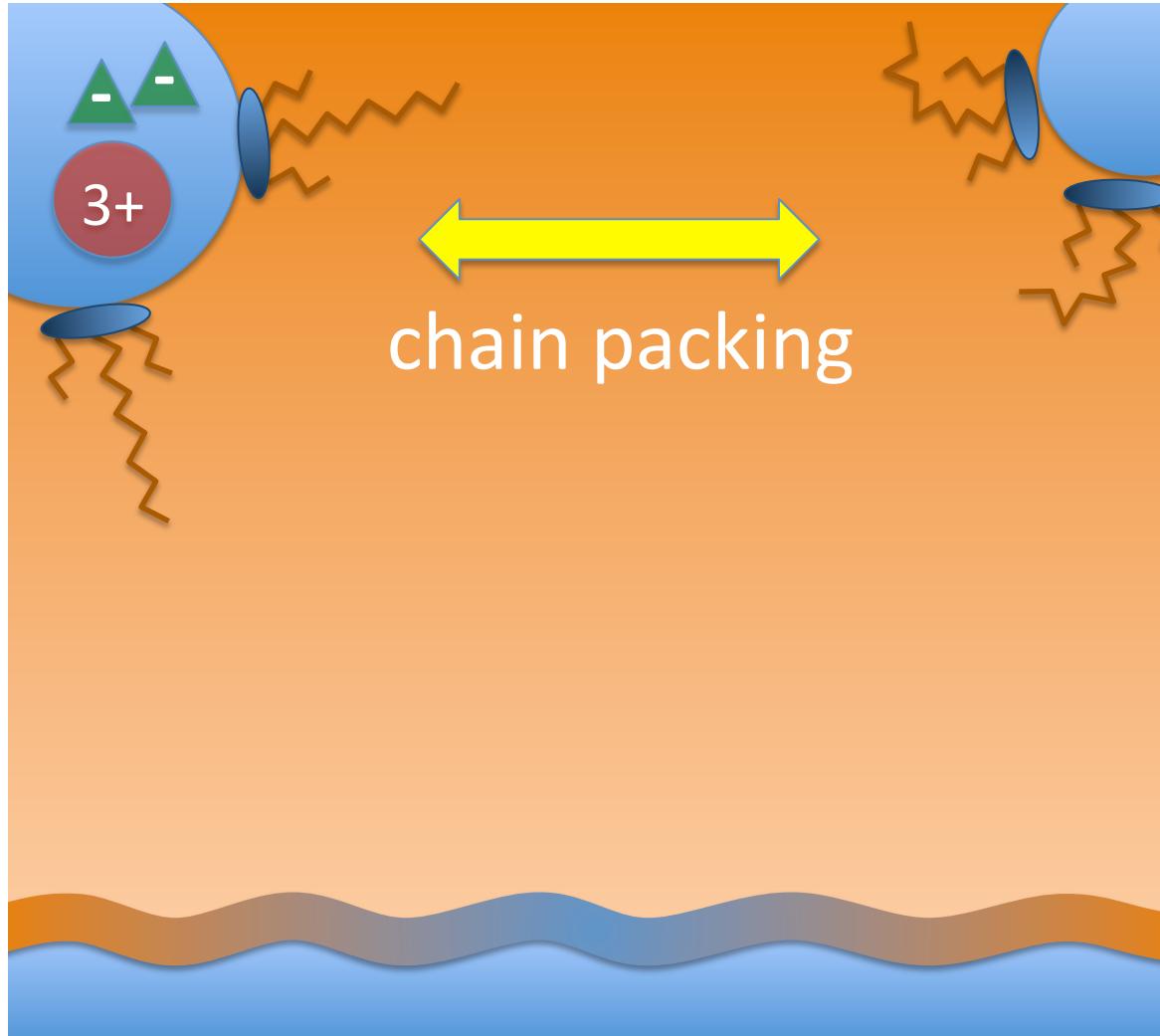
Hydrated ions and aggregates are chemical « complexes »  
but also « small » colloids

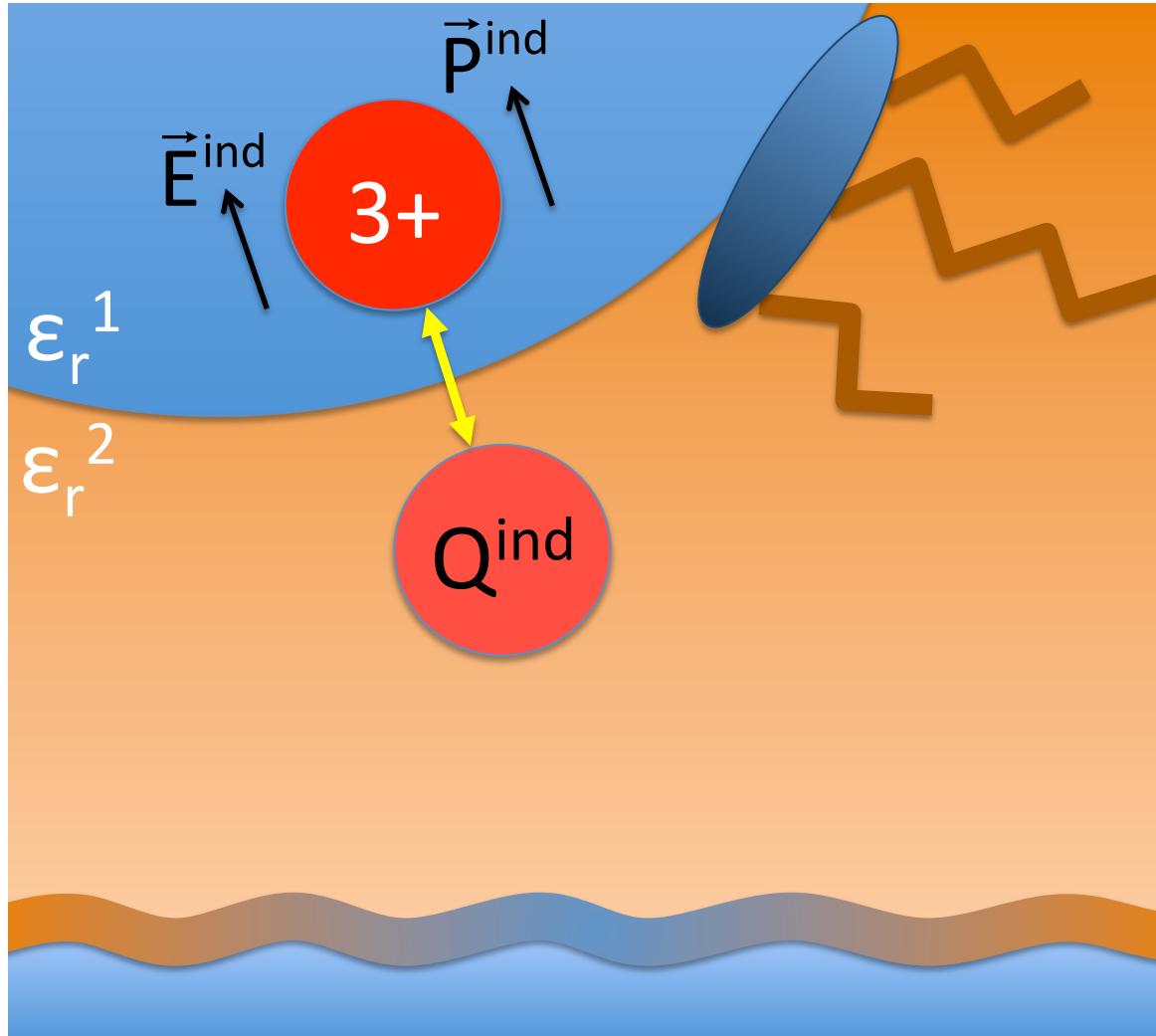
# Extractants in oil aggregate :





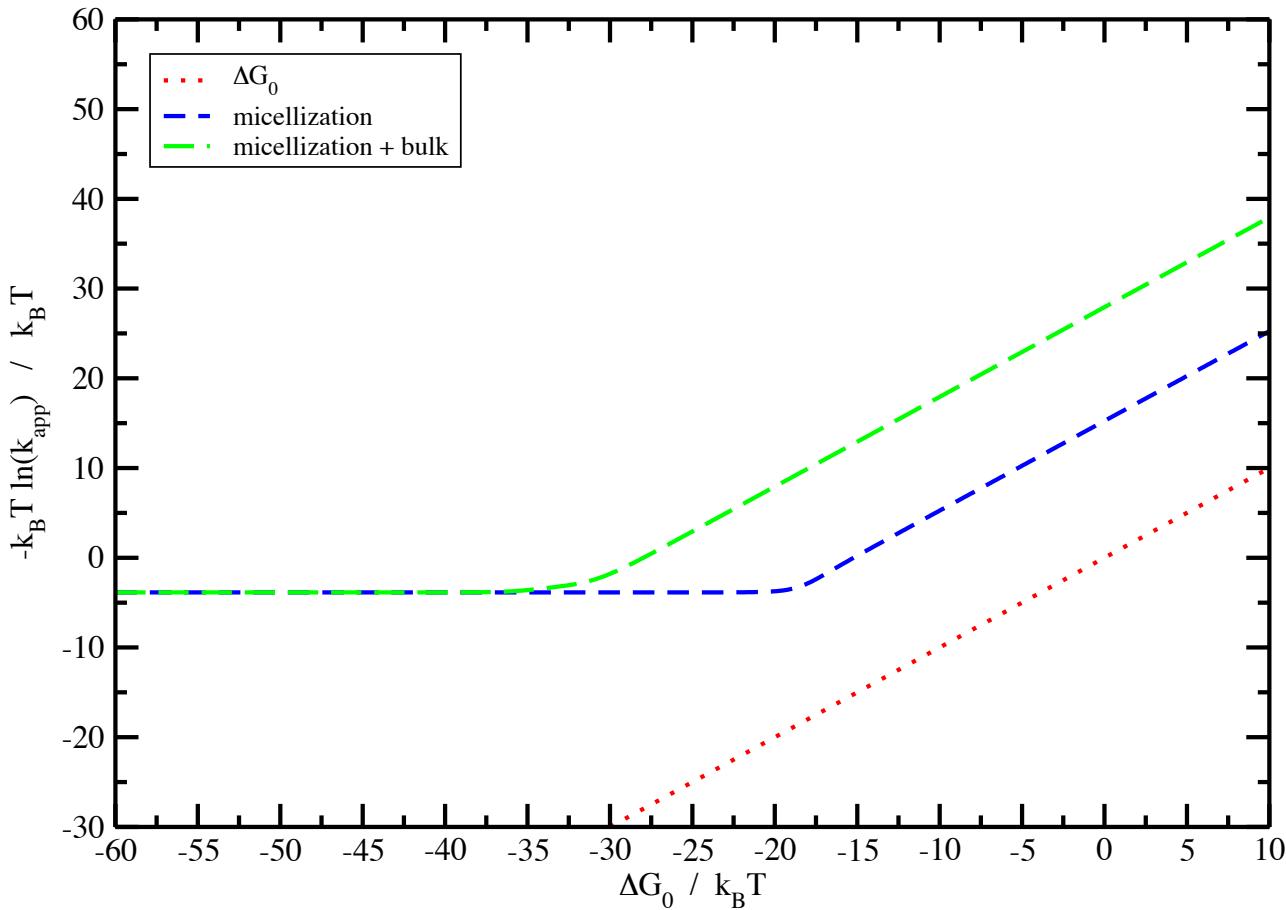
*Hydrated ions and aggregates are chemical « complexes »  
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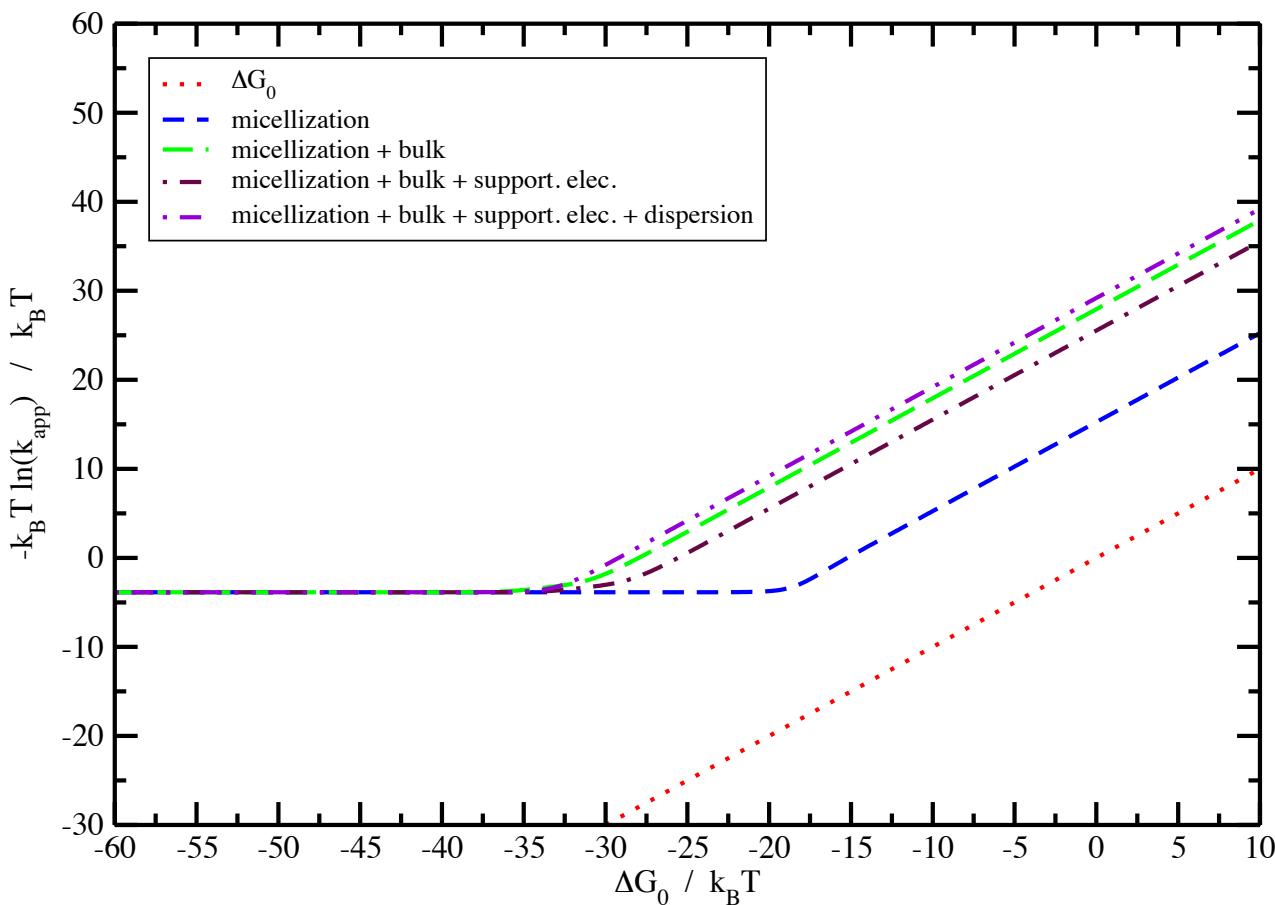
# *Ion pairs do not « like » oil !*

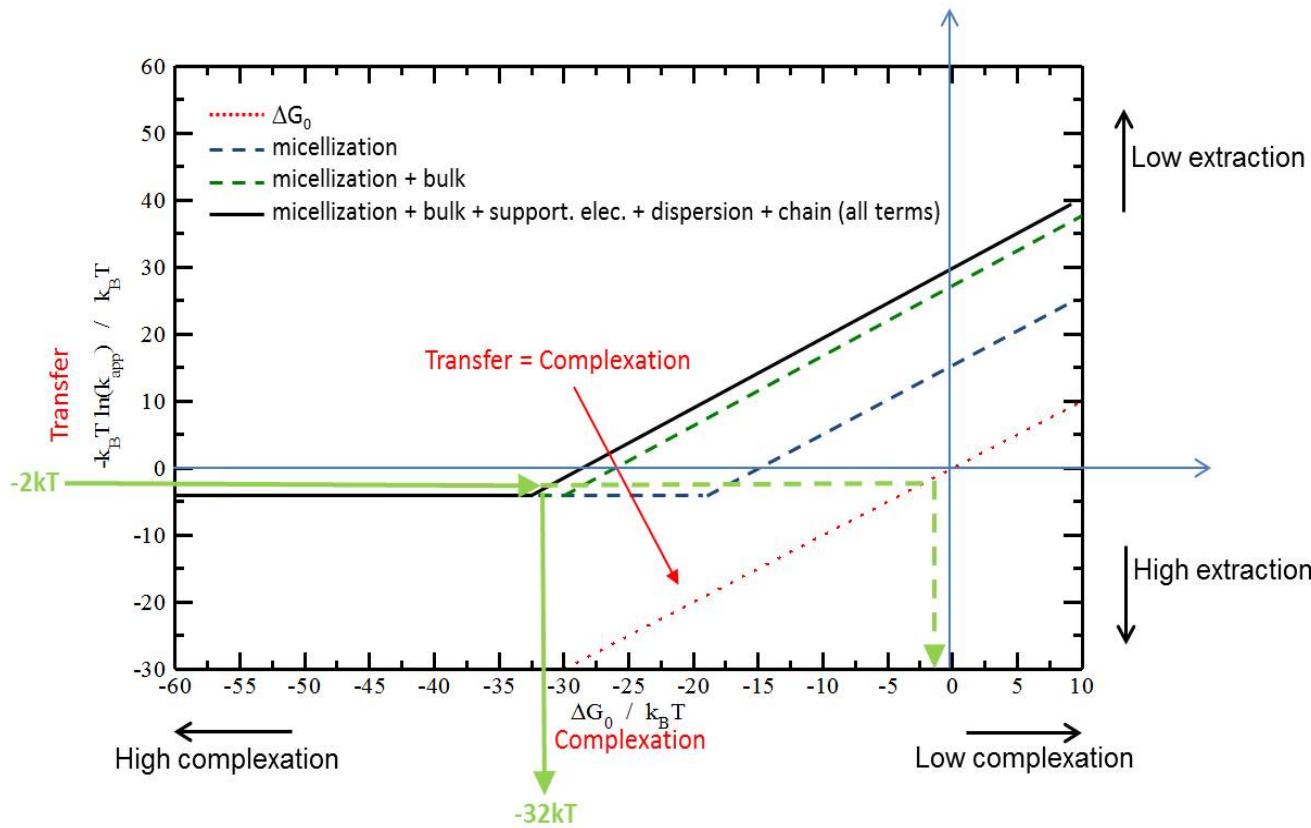


*See W. Kunz in « Specific ion effects », 2010*



## Dispersion/polarisation influence also :





*..Explains how complexation  $30 kT$  ( $70 \text{ kJ/mole}$ ) express in  $2 kT$ , ie  $5 \text{ kJ/Mole}$  extraction free energy « motor » in common industrial processes...*